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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/618,044	07/17/2000	Werner Pompetzki	9350-0169-0	7157

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EXAMINER

PRICE, ELVIS O

ART UNIT PAPER NUMBER

1621

DATE MAILED: 01/22/2002

H

Please find below and/or attached an Office communication concerning this application or proceeding.

Office Action Summary

Application No.

09/618,044

Applicant(s)

POMPETZKI ET AL.

Examiner

Elvis O. Price

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-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE ____ MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133).
- Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 29 November 2001.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-3 and 5-17 is/are pending in the application.
- 4a) Of the above claim(s) ____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) ____ is/are allowed.
- 6) ☒ Claim(s) 1-3 and 5-17 is/are rejected.
- 7) ☐ Claim(s) ____ is/are objected to.
- 8) ☐ Claim(s) ____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on ____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
- Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
- 11) ☐ The proposed drawing correction filed on ____ is: a) ☐ approved b) ☐ disapproved by the Examiner.
- If approved, corrected drawings are required in reply to this Office action.
- 12) ☐ The oath or declaration is objected to by the Examiner.

Priority under 35 U.S.C. §§ 119 and 120

- 13) ☒ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☒ All b) ☐ Some * c) ☐ None of:
1. ☒ Certified copies of the priority documents have been received.
2. ☐ Certified copies of the priority documents have been received in Application No. ____.
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).
- * See the attached detailed Office action for a list of the certified copies not received.
- 14) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. § 119(e) (to a provisional application).
- a) ☐ The translation of the foreign language provisional application has been received.
- 15) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. §§ 120 and/or 121.

Attachment(s)

- 1) ☒ Notice of References Cited (PTO-892) 4) ☐ Interview Summary (PTO-413) Paper No(s). ____
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948) 5) ☐ Notice of Informal Patent Application (PTO-152)
- 3) ☐ Information Disclosure Statement(s) (PTO-1449) Paper No(s) ____ 6) ☐ Other:

DETAILED ACTION

1. Claims 1-3 and 5-17 are pending in the application.
2. The 35 USC 112, first paragraph rejection of claims 1-3, 5, 6 and 9-15 has been withdrawn because applicants' arguments, filed 11/29/01, were found persuasive.

Priority

Receipt is acknowledged of papers submitted under 35 U.S.C. 119(a)-(d), which papers have been placed of record in the file.

Information Disclosure Statement

The information disclosure statement filed complies with the provisions of 37 CFR 1.97, 1.98 and MPEP § 609. It has been placed in the application file and the information referred to therein has been considered as to the merits.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

Claims 1-3 and 5-17 are rejected under 35 U.S.C. 103(a) as being unpatentable over Fukuhara et al. {U.S. Pat. 5,081,321}, in view of Hiles et al. {U.S. Pat. 4,626,604}.

Applicants claim a process for the hydrogenation of acetone, which comprises: conducting the liquid-phase hydrogenation of acetone having a water content of less than or equal to 1.0% by weight in at least two hydrogenation process stages, thereby preparing isopropanol product.

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Fukuhara et al. teach a process for the hydrogenation of acetone, which comprises conducting the liquid-phase hydrogenation of acetone in a reactor to produce isopropanol, in which a nickel containing catalyst supported on neutral alumina may be utilized as the hydrogenation catalyst and the reaction temperature is from room temperature to 200⁰ C and reaction pressure is from 2 to 50 bar (Col. 2, lines 20-40). The conversion and the yield of the isopropanol is 99.9%, respectively (see Example 1). The difference between applicants' claimed invention and the Fukuhara et al. invention is that the Fukuhara et al. reference is silent about multiple hydrogenation stages, the percentage of water contained in the acetone substrate and the percentage of by-products, if any.

Hiles et al. generally teach the unsaturated organic compounds, including acetone, can be hydrogenated to the corresponding product (isopropanol from acetone) utilizing multi-stage hydrogenation reactions (Col. 5, lines 14-45 and Col. 7, lines 50-59). Hiles et al. employ the multi-stage hydrogenation process to affect a greater conversion of the unsaturated organic compound to be hydrogenated to the corresponding product (see abstract)

It would have been *prima facie* obvious to one of ordinary skill in the art to prepare isopropanol by hydrogenating acetone as presently claimed, because Fukuhara et al. teach that acetone can be hydrogenated, in the liquid-phase, to produce isopropanol and Hiles et al. teach that compounds such as acetone can be hydrogenated in a multi-stage hydrogenation process. Since Fukuhara et al. are silent

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about any water contained in the acetone to be hydrogenated, the Examiner assumes that it is the same as presently claimed, absent evidence to the contrary.

One of ordinary skill in the art desiring to prepare highly pure isopropanol (more pure than that cited in the Fukuhara et al. reference) would have been motivated, in view of the Hiles et al. teachings, to incorporate additional hydrogenation stages, so as to optimize the total conversion of the isopropanol product, because Hiles et al. teach a multi-stage hydrogenation process, in which an unsaturated organic compound such as acetone is hydrogenated, can affect a greater conversion of the acetone to the corresponding hydrogenated product. The instantly claimed process would therefore have been obvious to one of ordinary skill in the art.

Response to Arguments

Applicant's arguments filed 11/29/01 have been fully considered but they are not persuasive.

Applicants argue that the Fukuhara et al. reference entails entirely different process features, which are critical to the process of Fukuhara et al. but are irrelevant as far as the process of the presently claimed invention is concerned, from the presently claimed invention in which the important features of the hydrogenation process are that the acetone reactant contain no more than 1% by weight water and that the hydrogenation process be conducted in at least two hydrogenation stages.

This argument is not convincing because the Fukuhara et al. process comprises the liquid-phase hydrogenation of acetone to produce isopropanol and applicants claims comprise a liquid-phase hydrogenation of acetone having a water content of less than

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or equal to 1.0% by weight in at least two hydrogenation process stages, thereby preparing isopropanol. The Examiner has asserted in the previous office action that, even though the Fukuhara et al. reference is silent about the water content in the acetone, there is no reason for the Examiner to believe that the Fukuhara et al. process does not utilize acetone having a water content of less than or equal to 1.0% by weight considering the high conversion and yield of the isopropanol product. Thus, the Examiner has assumed, based on the results of the Fukuhara et al. process, that the Fukuhara et al. reference is using acetone having a water content of less than or equal to 1.0% by weight. The Examiner would also like to call to applicants' attention that commercially available acetone contains less than 1.0% by weight as exemplified in the Sigma Catalog (see pg. 1681 of 1994 catalog).

The Examiner applied the Hiles et al. reference to address the multi-stage (at least two) hydrogenation limitation that was not taught in the Fukuhara et al. hydrogenation process. Since applicants have claimed at least two hydrogenation process stages in the hydrogenation of the presently claimed acetone, then the Hiles et al. reference is considered by the Examiner to be pertinent, analogous prior art which in fact teaches that it is advantageous to implement a multi-stage hydrogenation of unsaturated organic compounds, such as acetone, to produce the corresponding hydrogenated product with reduced byproduct formation. The skilled artisan then would be motivated, in view of the Fukuhara et al. and Hiles reference, to prepare isopropanol as presently claimed by purchasing commercially available acetone and hydrogenate

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conversion of the acetone and yield of the isopropanol product while reducing byproduct formation.

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Elvis O. Price whose telephone number is 703 605-1204. The examiner can normally be reached on 8:30 am to 5:00 pm; Mon-Fri.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Johann R. Richter can be reached on 703 308-4532. The fax phone numbers for the organization where this application or proceeding is assigned is 703 308-4556 for regular communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is 703 308-

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January 15, 2002


SAMUEL BARTS
PRIMARY EXAMINER
GROUP 1200